**H2O 2 Hydrogen and 1 Oxygen**



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**Total elements: 2**

**Total atoms: 3**

**Melting point: 0 degrees Celsius or 32 degrees Fahrenheit**

**Boiling point: 100 degrees Celsius or 212 degrees Fahrenheit**

**\*melting point and freezing points are the same**

**Facts about water**

Water is the only common compound that exists in solid, liquid, and gas phase under ordinary, natural conditions.

[Cohesion is](https://www.thoughtco.com/definition-of-cohesion-604933) a key property of water. Because of the polarity of the molecules, water molecules are attracted to each other.

Water may be called [the universal solvent](https://www.thoughtco.com/why-is-water-the-universal-solvent-609417) because it is able to dissolve many different substances.

Ice is less dense than liquid water.